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The non-essentiality of the N-terminal amino group of papain

Studies concerning the pH dependence of the hydrolysis of a number of specific substrates by papain have shown that on the alkaline side of the pH optimum (pH 6), K_m is governed by a protein group with a pK of 8.5–9.0 (refs. 1, 2). In chymotrypsin a group of similar pK with a similar function has been identified as the α -amino group of N-terminal isoleucine³. Therefore it was considered worthwhile to find out whether or not the N-terminal isoleucine is involved in substrate binding in papain, too. It will be shown in this communication that this is not the case.

Guanidination⁴ or acetylation⁵ of most of the amino groups does not affect the activity of papain, but it was not proved in these experiments that the α -NH₂ was modified.

Therefore, in the present investigation, papain was treated with 1-fluoro-2,4-dinitrobenzene (FDNB) and the product was tested for modification of the N-terminal isoleucine group and for activity.

Crystals (modification C) of 30 mg of mercuri-papain, were suspended in 10 ml of 60% (v/v) of methanol, 15 mM in bicarbonate and 5 mM in FDNB. After storage for one night at 20° the crystals were washed 3 times with 60% (v/v) methanol and brought to the initial volume for assay. Since the modified crystals proved to be insoluble in water, all activities were determined in the crystalline state towards 0.05 M acetylglycine ethyl ester in 20% of Na₂SO₄ (ref. 6). The activity of DNP-papain was found to be the same as that of control crystals.

The remaining portion of DNP-papain was washed with 60% methanol, methanol and ether. The dry preparation (20 mg) was hydrolyzed (16 h in 6 M HCl at 105°). The hydrolysate was taken up in ethanol and analyzed on thin-layer plates (Kieselgel F 254, Merck), run with propanol–ammonia (7:3, v/v) together with reference DNP-compounds. The spots of DNP-isoleucine were scraped off and eluted with ethanol. The absorbance of the eluates was read at 348 nm.

The DNP-isoleucine content was found to be 96 \pm 5% of the calculated value. Furthermore 4.5 moles of $\epsilon\text{-DNP-lysine}$ and about 3 moles of O-DNP-tyrosine per

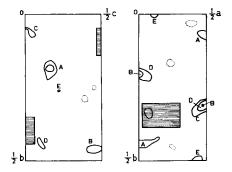


Fig. 1. Difference Fourier maps, the Okl and hkO projection two-dimensional space group pgg, of DNP-papain versus papain. Contour lines drawn at a level significantly above the normal background noise. Holes in the projections are surrounded by a dotted line. The shaded area indicates the active center.

SHORT COMMUNICATIONS 393

mole of protein were estimated in the ether-extracted hydrolysate from the absorbance at 350 and 390 nm in I M HCl (cf. Table II-I of ref. 7).

X-ray patterns of crystals treated in the same way were made and compared with those of unmodified crystals. Two difference Fourier maps, the Okl and hkO projections, are shown in Fig. 1. Although the electron density peaks do not rise much above the background noise, six of them can be clearly seen. Peaks B, C, D overlap in the hkO projection. Peak A, elongated in the hkO projection, can most likely be ascribed to two partially overlapping peaks; this overlap also occurs in the other projection, as Peak A here has a rather high value. By correlation of the y-coordinates in the two projections, three-dimensional coordinates of the peaks can be calculated, thus determining which amino acid side chains have reacted.

Peak A can be ascribed to reaction with the α-NH₂ group and Lys-137, Peak B with Lys-54, Peak C with Tyr-176, Peak D with Tyr-161, and Peak E with Tyr-183. No reaction with Cys-25 or His-106 in the active site could be seen. Of course, sites of minor reactivity do not show up.

The present data exclude the participation of the α -NH₂ group in the action of papain, which apparently differs from chymotrypsin in this respect. Furthermore a number of other groups are not essential for catalysis either.

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Allosteric effects and phlorizin inhibition of intestinal trehalase

Intestinal sucrase (sucrose glucohydrolase)1,2, and isomaltase (isomaltose glucohydrolase)3, are activated by Na+, and at least in the case of sucrase, the kinetic parameters of this Na⁺ activation are essentially the same as those of Na⁺ activation of sugar transport, although sucrase itself does not participate in sugar transport^{4,5}. Sucrase shows an evident cooperative interaction both among substrate sites and among Na⁺ sites⁶.